

Tetrahedron Letters, Vol. 38, No. 39, pp. 6825-6828, 1997 © 1997 Elsevier Science Ltd All rights reserved. Printed in Great Britain 0040-4039/97 \$17.00 + 0.00

PII: S0040-4039(97)01605-5

Total Synthesis of Rhizoxin D

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Abstract: The convergent, enantiocontrolled synthesis of a significant antimitotic agent for cancer chemotherapy is presented with completion of rhizoxin diene 2.
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The rhizoxins are a family of sixteen-membered macrolactones with exceptionally potent antitumor and antifungal activity. Produced by *Rhizopus chinensis*, rhizoxin (1) was first discovered as the causal agent of rice seedling blight.¹ Subsequent explorations² have led to characterizations of several related structures, including the macrocyclic diene 2 (rhizoxin D),³ which is a putative precursor to the *bis*-epoxide 1. Studies revealing the absolute configuration of 1 and a biosynthetic pathway have been reported.⁴



The rhizoxins exhibit powerful antimitotic effects as reversible inhibitors of the polymerization of intracellular α - and β -tubulin to prevent microtubule formation.⁵ One striking feature is that the rhizoxins exhibit no cross-resistance to agents such as cisplatin, doxorubicin, and etoposide.⁶ Rhizoxin and maytansine share a related binding site, which is non-identical to the receptor for other inhibitors, including colchicine, vinblastine, dolastatins, and halichondrin.⁷ Limited structure-activity information suggests that **2** exhibits equivalent potency to rhizoxin itself.^{5,6a} *In vivo* activity has been demonstrated for solid tumors, including five human tumor xenografts: LOX melanoma, MX-1 breast cancer, non-small cell lung cancer A549, and small cell lung cancers LXFS605 and LXFS650.⁸ Phase II clinical studies of rhizoxin have been completed, and Phase III evaluations are underway. In 1993, Ohno and coworkers⁹ communicated the first synthesis of rhizoxin. Recently, Kende has reported a route for the preparation of **2**,¹⁰ and studies have described various structural fragments.¹¹ Herein we present our efforts culminating in the enantiocontrolled total synthesis of rhizoxin D (**2**).

From the onset, our plans were directed with two underlying assumptions. Our experiences with macrolactonizations suggested that ring closure of a hindered C-15 alcohol with an activated acyl derivative of the stabilized α , β -unsaturated acid (C₁) would be difficult compared to the intramolecular Horner-Emmons process envisioned from a C-2/C-3 disconnection. Secondly, this strategy introduced an element of pseudosymmetry with a C-3 aldehyde and the carbonyl of the valerolactone arising from a two-carbon branch at C-5 of **2**. These concepts were realized beginning with 3-cyclopentenylacetaldehyde¹² for preparation of the C₃-C₁₂ fragment

(Scheme I). Using an Evans aldol reaction,¹³ the boron Z(O) enolate derived from the (S)-4-benzyloxazolidinone 4 provided exclusive formation of the *syn*-adduct securing the asymmetry at C_7 and C_8 of 5. Scheme I^a



^aKey: a) *n*-Bu₂BOTf, Et₃N, CH₂Cl₂, 0 °C to -78 °C then 3, 95%; b) DHP, PPTs, CH₂Cl₂, 100%; c) OsO₄, NMO, Acetone/ H₂O, 88%; d) cyclopentanone, PPTs, HC(OEt)₃, CH₂Cl₂, 92%; e) LiBH₄, MeOH, Et₂O, 0 °C, 91%; f) TsCl, Et₃N, CH₂Cl₂, 98%; g) ArSH, NaH, DMF/THF, 0 °C, 95%; h) *m*CPBA, NaHCO₃, CH₂Cl₂, 98%; i)LDA, HMPA, THF, -78 °C then (*E*)-1,3dibromo-2-butene, 88%; j) DBU, tol., 105 °C, 70%.

Dihydroxylation of **5b** afforded a mixture (7:1 ratio) of cis-diols.¹⁴ Diastereofacial selectivity of the oxidation was only evident from ¹³C-NMR spectra of crude triols obtained via removal of the C₇ THP ether. In all other aspects, the compounds of the synthesis route behaved as simple (1:1 ratios) THP isomers. Ketalization of the diols from **5** followed by hydride reductive cleavage¹⁵ of the chiral auxiliary yielded alcohol **6**. Standard transformations gave **7**, and deprotonation to the corresponding α -sulfonyl carbanion led cleanly to C-alkylation.¹⁶ Subsequent E₂ elimination of the intermediate allylic sulfone produced *trans*-diene **9** in 70% isolated yield. In contrast, use of the *para*-tolylsulfone **8** led to only 30% yield of **9** with evidence of decomposition at higher temperatures. Base-induced sulfone elimination with KO^tBu (1 <u>M</u> in THF) at 22 °C afforded the analogous (*E*)-enyne (80% yield) via the further loss of HBr from **9**.

Synthesis of the $C_{13} \rightarrow C_{20}$ fragment was accomplished on a multigram scale (Scheme II) by initially establishing the C_{16} and C_{17} asymmetry with high enantiocontrol via the Evans aldol condensation^{13,15} with α,β -unsaturated aldehyde 10. The remaining stereocenter at C_{15} was installed by the Sharpless asymmetric Scheme II^a



^aKey: a) *π*-Bu₂BOTf, Et₃N, CH₂Cl₂, 0 °C to -78 °C then **10**, 87%; b) Bu₃B, LiBH₄, HOAc, THF, 77%; c) TBDMSCI, Imid., CH₂Cl₂, (100%); d) *i*. NaH, CH₃I, THF, *ii*. PPTs, EtOH, 66 °C, 97% (two steps); e) (COCI)₂, DMSO, Et₃N, CH₂Cl₂, -78 °C to 22 °C, then (C₆H₅)₃PCHCO₂CH₃, 88%; f) DIBAL, CH₂Cl₂, 93%; g) Ti(OⁱPr)₄, (-)-DET, ⁱBuOOH, CaH₂, 4Å sieves, CH₂Cl₂, 93%; h) Red-AI, THF, -78 °C to 22 °C, 90%; i) Piv-CI, pyr, CH₂Cl₂, 90%; j) Me₂ⁱBuSiOTf, collidine, CH₂Cl₂, 98%; k) DIBAL, CH₂Cl₂, -78 °C to 22 °C, 92%.

epoxidation (88% de) of *E*-allylic alcohol 13 provided from the Wittig chain extension of primary alcohol 12. Regioselective hydride opening¹⁷ of the epoxyalcohol from 13 gave diol 14 for transformation into β -silyloxyaldehyde 15. The choice of the silyl protection at C₁₅ was important for steric shielding to prevent base-induced α -deprotonation in 15. This was apparent in the BOM ether and the analogous C₁₅ benzoate of 15, which suffered from β -eliminations in the Swern oxidation or in further attempts for nucleophilic additions.

Linkage of our nonracemic fragments was achieved through halogen-metal exchange of 9 (¹BuLi, 2 equiv., THF/Et₂O/pentane (4:1:1) at -120 °C; warming to -90 °C) followed by addition of aldehyde **15** (-90 °C to -78 °C). The process afforded a mixture (1:1 ratio) of C_{13} alcohol diastereomers **16** in 77% yield (Scheme III).¹⁸ Undesired isomer **16b** was readily separated by flash chromatography and effectively converted into **16a** via TPAP oxidation¹⁹ and borohydride reduction using (*R*)-2-methyl-CBS-oxazaborolidine,²⁰ giving a 68% overall yield of the desired C_{13} alcohol. Interestingly, the conjugated enone system displayed an inherent facial bias upon reaction with simple hydrides (LiAlH₄) for highly selective production of **16b**. After C_{13} hydroxyl protection and fluoride treatment, esterification with diisopropylphosphonoacetic acid was followed by



^{*a*}Key: a) TPAP, NMO, 4Å sieves, CH₂Cl₂, 80%; b) (*H*)-2-methyl-CBS-oxazaborolidine, BH₃, THF, -10 °C, 85%; c) SEMCl, ⁱPr₂EtN, CH₂Cl₂, 94%; d) TBAF, THF, 94%; e) diisopropylphosphonoacetic acid, 1-cyclohexyl-3-(2-morpholinoethyl)carbodiimide methyl-*p*-toluenesulfonate, DMAP, 4Å sieves, CH₂Cl₂, 97%; f) HOAc/H₂O (4/1), 48 h, 56%; g) NalO₄, THF/H₂O; h) TPAP, NMO, 4Å sieves, CH₂Cl₂, 67%, 2 steps; i) DBU (2 equiv), CH₃CN, LiCl (15 equiv), 4.6 x 10⁻⁴ M, 84%; j) DDQ, H₂O, CH₂Cl₂, 98%; k) MnO₂, 70%; l) KHMDS, 4-[3-(diphenylphosphinoyl)-2-methylpropenyl]-2-methyloxazole, 44%; m) Me₂BBr, THF/CH₂Cl₂, 0 °C, 65%

simultaneous hydrolysis of the THP ethers and cyclopentylidene ketal to afford the key intermediate triol 17.²¹ Oxidative cleavage of the vicinal diol provided two diastereomeric lactols (ratio 3:2). Development of stereochemistry at C₅ in the lactols was based on a clear thermodynamic preference for diequatorial (C₅/C₇) substitution. These lactols underwent a mild TPAP oxidation to a single lactone 18 in 67% overall yield. Finally, intramolecular Horner-Emmons cyclization²² of 18 was achieved in 84% yield necessitating strictly anhydrous conditions to avoid hydrolysis of the δ -lactone in 19. Quantitative DDQ removal²³ of the *para*methoxybenzyl ether produced the parent rhizoxin macrocycle 20. Completion of the total synthesis of **2** was realized by oxidation of the allylic alcohol **20**, and Wittig olefination (-78 °C, THF) with the carbanion generated from 4-[3-(diphenylphosphinoyl)-2-methyl-propenyl]-2-methyloxazole.⁹ The all-*trans*-triene was obtained in 44% yield,²⁴ and deprotection of the C₁₃ SEM ether afforded synthetic rhizoxin D, $[\alpha]_D^{24} + 293^\circ$ (c 0.53, MeOH), with characterization data which were identical to the literature for the natural product.^{2a}

Acknowledgment: We thank the National Institutes of Health (GM42897) for generous financial support. A grant from Merck Research Laboratories is also gratefully acknowledged.

REFERENCES

- Tsuruo, T.; Oh-hara, T.; Iida, H.; Tsukagoshi, S.; Sato, Z.; Matsuda, I.; Iwasaki, S.; Okuda, S.; Shimizu, F.; Sasagawa, K.; Fukami, M.; Fukuda, K.; Arakawa, M. Cancer Res. 1986, 46, 381. Iwasaki, S.; Kobayashi, H.; Furukawa, J.; Namikoshi, M.; Okuda, S.; Sato, Z.; Matsuda, I.; Noda. T. J. Antibiot. 1984, 37, 354.
- a) Iwasaki, S.; Namikoshi, M.; Kobayashi, H.; Furukawa, J.; Okuda, S. Chem. Pharm. Bull. 1986, 34, 1387. b) Kiyoto,
 S.; Kawai, Y.; Kawakita, T.; Kino, E.; Okuhara, M.; Uchida, I.; Tanaka, H.; Hashimoto, M.; Terano, H.; Kohsaka, M.;
 Aoki, H.; Imanaka, H. J. Antibiot. 1986, 39, 762.
- 3. Although the rhizoxin diene 2 was not named by investigators responsible for its discovery (ref. 2a), we have referred to this natural product as rhizoxin D. An account of our total synthesis was previously presented: Williams, D.R.; Werner, K.M.; Feng, B. National Meeting; American Chemical Society, 1996, 211th, ORGN 442 (New Orleans).
- a) Iwasaki, S.; Namikoshi, M.; Kobayashi, H.; Furukawa, J.; Okuda, S.; Itai, A.; Kasuya, A.; Iitaka, Y.; Sato, Z. J. Antibiot. 1986, 39, 424. b) Kobayashi, H.; Iwasaki, S.; Yamada, E.; Okuda, S. Chem. Commun. 1986, 1702.
- 5. Takashi, M.; Iwasaki, S.; Kobayashi, H.; Okuda, S.; Murai, T.; Sato, Y.; Haraguchi-Hiraoka, T.; Nagano, H. J. Antibiot. 1987, 40, 66.
- a) Kato, Y.; Ogawa, Y.; Imada, T.; Iwasaki, S.; Shimazaki, N.; Kobayashi, T.; Komai, T. J. Antibiot. 1991, 44, 66. b) Takigawa, N.; Ohnoshi, T.; Ueoka, H.; Horiguchi, T. Gan. To. Kagaku Ryoho, 1993, 20, 1221.
- a) Takahashi, M.; Iwasaki, S.; Kobayashi, H.; Okuda, S.; Murai, T.; Sato, Y. Biochim. Biophys. Acta, 1987, 926, 215. b) Hamel, E. Pharmac. Ther. 1992, 55, 31. c) Recent investigations have demonstrated that the 100th amino acid (Asn-100) of β-tubulin is essential for rhizoxin-binding: Li, Y.; Kobayashi, H.; Hashimoto, Y.; Iwasaki, S. Biochem. Biophys. Res. Commun. 1992, 187, 722.
- 8. Hendriks, H.R.; Plowman, J.; Berger, D.P.; Paull, K.D. Ann. Oncol. 1992, 3, 755.
- a) Nakada, M.; Kobayashi, S.; Iwasaki, S.; Ohno, M. Tetrahedron Lett. 1993, 34, 1035. b) Nakada, M.; Kobayashi, S.; Shibasaki, M.; Iwasaki, S.; Ohno, M. Tetrahedron Lett. 1993, 34, 1039.
- 10. Kende, A.S.; Blass, B.E.; Henry, J.R. Tetrahedron Lett. 1995, 36, 4741.
- For recent efforts: a) Kobayashi, S.; Nakada, M.; Ohno, M. Pure & Appl. Chem. 1992, 64, 1121. b) Lafontaine J.A.; Leahy, J.W. Tetrahedron Lett. 1995, 36, 6029. c) Provencal, D.P.; Gardelli, C.; Lafontaine, J.A.; Leahy, J.W. Tetrahedron Lett. 1995, 36, 6033. d) Keck, G.E.; Savin, K.A.; Weglarz, M.A.; Cressman, E.N.K. Tetrahedron Lett. 1996, 37, 3291. See also citations in reference 10.
- 12. Wilt, J.W.; Massie, S.N.; Dabek, R.B. J. Org. Chem. 1970, 35, 2803.
- 13. a) Gage, J.R.; Evans, D.A. Org. Syn. 1989, 68, 77. b) Gage, J.R.; Evans, D.A. Ibid. 1989, 68, 83.
- 14. For a related example: McMurry, J.E.; Dushin, R.G. J. Am. Chem. Soc. 1990, 112, 6942.
- 15. Evans, D.A.; Sjogren, E.B.; Bartroli, J.; Dow, R.L. Tetrahedron Lett. 1986, 27, 4957.
- For stereocontrolled preparation of E-3-bromo-2-buten-1-ol: Schlosser, M.; Hammer, E. Helv. Chim. Acta, 1974, 57, 2547. The alcohol was converted to the corresponding bromide with NBS/Ph3P at 0 °C.
- 17. Finan, J.M.; Kishi, Y. Tetrahedron Lett. 1982, 23, 2719.
- Stereochemistry was established from comparisons of ¹³C-NMR data for the pair of acetonides individually characterized from desilylation of 16 Sec: Rychnovsky, S.D.; Skalitzky, D. J. Tetrahedron Lett. 1990, 31, 945.
- 19. Ley, S.V.; Norman, J.; Griffith, W.P. Marsden, S.P. Synthesis, 1994, 639.
- Corey, E.J.; Bakshi, R.K.; Shibata, S.; Chen, C.-P.; Singh, V.K. J. Am. Chem. Soc. 1987, 109, 7925. Mathre, D.J.; Jones, T.K.; Xavier, L.C.; Blacklock, T.J.; Reamer, R.A.; Mohan, J.J.; Jones, E.T.T.; Hoogsteen, K.; Baum, M.W.; Grabowski, E.J.J. J. Org. Chem. 1991, 56, 751.
- 21. Reactions also provided a 15% yield of the C₇ alcohol resulting only from removal of the THP ether. However, longer reaction times or increased temperatures led to numerous side products.
- 22. For an example: Williams, D.R.; McGill, J.M. J. Org. Chem. 1990, 55, 3457, and references therein.
- 23. Horita, K.; Yoshioka, T.; Tanaka, T.; Oikawa, Y.; Yonemitsu, O. Tetrahedron, 1986, 42, 3021.
- 24. In practice, our Wittig reactions led to crude product with some hydrolytic opening of the valerolactone. Submission to Yamaguchi conditions (procedure: Suzuki, K.; Tomooka, K.; Katayama, E.; Matsumoto, T.; Tsuchihashi, G. J. Am. Chem. Soc. 1986, 108, 5221) recovered this material prior to purification.

(Received in USA 22 July 1997; revised 31 July 1997; accepted 1 August 1997)